





# XVII ICC ISTANBUL 2022 25 - 29 JULY

New Interfaces Bridging Continents and Cultures with Clays



SCIENTIFIC RESEARCH



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## AIPEA – XVII INTERNATIONAL CLAY CONFERENCE ICC 2022

### 25–29 JULY 2022 HILTON MASLAK ISTANBUL, TURKEY

SCIENTIFIC RESEARCH ABSTRACTS







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#### The characteristics of V(V) and P(V) adsorption by LDH derived from magnesite: kinetics, pH influence and competition with common anions

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With a vast amount of research devoted to new adsorbents for wastewater treatment, special attention is given to the development of simple low-cost materials. This study described the adsorption properties of Mg/Al and Mg/Fe LDH derived from magnesite (M) (Rybka et al., 2022). The affinity towards V(V) and P(V) of the studied materials was compared with that of LDH conventionally obtained from chemical reagents.

The structure of LDH is built of brucite-like layers build from divalent ( $M^{II}$ ) and trivalent ( $M^{III}$ ) metal octahedra. The positive charge of the layers is balanced by easily exchangeable hydrated anions. This makes LDH a promising material for the removal of anionic pollutants. In this work, the M was used as a  $M^{II}$  source substituting the conventionally used chemical reagents. It was dissolved in AlCl<sub>3</sub> or HCl to release Mg(II) for the synthesis of Mg/Al and Mg/Fe LDH, respectively. The LDH materials were obtained with a co-precipitation method.

The studied materials were compared to the reference LDH samples obtained exclusively from the chemical reagents. The maximum capacity of the materials was tested in the removal reactions of V(V) and P(V) chosen as toxic and eutrophogenic elements, respectively. The minimal doses of the adsorbents were determined for the effective removal of low initial concentrations ( $C_{in} = 0.05 \text{ mmol/L}$ ) of the anions. The kinetics, pH effect (pH = 3 and 5) and co-existing anions ( $SO_4^{2-}$ ,  $NO_3^{-}$ ) influence on adsorption efficiency were described.

The materials showed a high affinity for V(V) and P(V). The Fe-bearing materials were more efficient than their Al-containing analogues. The removal rate of P(V) was lower than that of V(V). The SO<sub>4</sub><sup>2-</sup> influenced the removal rate of V(V) by Al-bearing LDH. The lower pH increased the removal of P(V) by Fe-bearing materials. In the other cases, the adsorption efficiency of V(V) and P(V) did not decrease, which confirmed the high stability of the materials at pH = 3. In turn, the removal of competitive anions drastically decreased.

Rybka K., Matusik J., Marzec M. (2022) Mg/Al and Mg/Fe layered double hydroxides derived from magnesite and chemicals: The effect of adsorbent features and anions chemistry on their removal efficiency. *Journal of Cleaner Production*, 332, 130084 This project was supported by the National Science Centre Poland, under a research project awarded by Decision No. 2017/27/B/ST10/00898.